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Solution The average atomic weight of silicon (A) is computed by adding fraction-of-occurrence/atomic weight Si products for the three isotopes—i.e., using Equation 2.2. (Remember: fraction of occurrence is equal to the percent of occurrence divided by 100.) Thus, A Si = f 28 A 28 + f 29 A 29 f 30 A Si Si Si Si 30 Si

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Chapter 9: Phase Diagrams
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